

IN THE CLAIMS:

Please cancels 1-16.

Please add the following new claims:

1                   17.   (New) An electrochemical generator that may be used as a  
2 primary or secondary electrochemical generator, the electrochemical generator  
3 comprising two electrodes supporting different electro-active materials, the electrodes  
4 being connected together by an electrolyte, wherein the electro-active material used in the  
5 composition of at least one electrode includes one of an oxide or chalcogenide of  
6 transition metals, or their at least partially lithiated form.

1                   18.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the oxide or chalcogenide of transition metals, or their at least partially lithiated  
3 form, is selected from a group consisting of  $\text{TiO}_2$ ,  $\text{Nb}_2\text{O}_5$ ,  $\text{WO}_3$ ,  $\text{MnO}_2$ ,  $\text{HfO}_2$ ,  $\text{TiS}_2$ ,  $\text{WS}_2$ ,  
4  $\text{TiSe}_2$ ,  $\text{Li}_x\text{NiO}_2$ ,  $\text{Li}_x\text{CoO}_2$ ,  $\text{Li}_x(\text{NiCo})\text{O}_2$ ,  $\text{Fe}_2\text{O}_3$ ,  $\text{Fe}_3\text{O}_4$ ,  $\text{RuO}_x$ ,  $\text{Fe}_x\text{S}_2$ ,  $\text{Ru}_x\text{S}_2$ ,  $\text{MoS}_2$ ,  $\text{WS}_2$ ,  
5  $\text{Ir}_x\text{O}_2$ ,  $\text{Ce}_x\text{O}_2$ ,  $\text{Li}_x\text{Na}_y\text{MnO}_2\text{I}_n$  ( $n < 1$ ),  $\text{In}_x\text{O}_3$ ,  $\text{Ta}_x\text{O}_5$ ,  $\text{SnM}_x\text{O}_y$ , wherein M is one of a mixture  
6 of the elements in the group, or  $\text{Sn}_x\text{O}_2$  in mesoporous form having a pore size of 0.001 to  
7 10 micrometers (micron) and a specific surface area between 2 and 2000  $\text{m}^2/\text{g}$ .

1                   19.   (New) An electrochemical generator in accordance with claim 17  
2 wherein at least one electrode is a mesoporous electrode, and wherein the mesoporous  
3 electrode has a three dimensional bicontinuous structure consisting of an interconnected  
4 solid phase material used for reversible ion intercalation and for electronic transport that  
5 is in contact with an interconnected porous space filled with electrolyte that serves for  
6 ionic transport.

1                   20.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is mesoporous and is prepared by precipitating a  
3 precursor compound of the electro-active material in an aqueous solution via a sol-gel

4 method followed by deposition of a precursor on a conductive support and sintering at a  
5 temperature between 300 and 800°C.

1                   21. (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is mesoporous and is prepared by chemical reaction  
3 between precursor compounds in aqueous solution in the presence of surfactant micelles,  
4 and wherein the surfactant micelles act as templates to produce a desired mesoporous  
5 morphology.

1                   22. (New) An electrochemical generator in accordance with claim 21  
2 wherein the desired mesoporous morphology is in the form of ordered hexagonal arrays.

1                   23. (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is composed of elongated interconnected members  
3 that are connected to other elongated members at at least two points of an aspect ratio of  
4 at least 4 and of which a small dimension of the member is smaller than 300 nm.

1                   24. (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is prepared by inclusion of solvents to a particle or  
3 precursor and thus exerts control over the texture and morphology of the electro-active  
4 material and of the porosity of the electrode, which may be controlled from 70% to 25%  
5 by changing the ratio of oxide precursor grains versus solvent.

1                   25. (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is mesoporous and is comprised of mesoporous beads  
3 or rods that are 5-20 micrometers (microns) that are electrically connected together by  
4 compressing, in the form of pellets or films, a mixture composed of the beads or rods, of  
5 carbon powder or conducting polymer and of a bonding material contained in solvent,  
6 and then performing drying the mixture on at least one of a conducting support and baking  
7 at a temperature for binder or conductive matrix conversion to a desired state.

1                   26.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the electrolyte contains alkali or alkaline earth metals in cationic form.

1                   27.   (New) An electrochemical generator in accordance with claim 26  
2 wherein the alkaline metal is lithium in the form of one of its salts chosen from a group  
3 consisting of tetrafluoroborate, hexafluorophosphate, hexafluoroantimonate,  
4 hexafluoroarsenate, trifluoromethane sulfonate, bis-(trifluorosulfonyl) imide, tris-  
5 (trifluorosulfonyl)methide, trifluoro-methanesulfonate, trifluoroacetate,  
6 tetrachloroaluminate and perfluorobutane sulfonate.

1                   28. An electrochemical generator in accordance with claim 17 wherein  
2 the electrolyte includes an aprotic solvent selected from a group consisting of ethylene  
3 carbonate, propylene carbonate, dimethylcarbonate, diethylcarbonate, dioxolane,  
4 butyrolactone, methoxypropionitrile, methoxy-ethoxy propionitrile, methoxy-  
5 diethoxypropionitrile, methoxyacetonitrile, tetrafluoro-propanol and combinations  
6 thereof.

1                   29.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the electrolyte includes a molten salt as a solvent for a lithium ion containing salt.

1                   30.   (New) An electrochemical generator in accordance with claim 29  
2 wherein the molten salt is selected from a group consisting of methyl-ethyl-imidazolium  
3 trifluoromethanesulfonate, methyl-ethyl-imidazolium bis (trifluorosulfonyl) imide and  
4 alkylguanidinium bis (trifluorosulfonyl) imide.

1                   31.   (New) An electrochemical generator in accordance with claim 26  
2 wherein the electro-active material of at least one of the electrodes is capable of forming  
3 an intercalation compound with the alkali or alkaline earth metal.

1                   32.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material of one electrode is composed of  $\text{TiO}_2$  in anatase form  
3 having a mesoporous structure.

1                   33.   (New) An electrochemical generator in accordance with claim 17  
2 wherein one of the two electrodes is a negative electrode and the other is a positive  
3 electrode, and the negative electrode includes mesoporous  $\text{TiO}_2$  and the composition of  
4 the positive electrode includes  $\text{Li}_y\text{Mn}_2\text{O}_4$  ( $y < 2$ ) in amorphous or crystalline form, and  
5 wherein the electrolyte is composed of a 1M solution of lithium bis-(trifluorosulfonyl)  
6 imide in methoxypropionitrile as a solvent.

1                   34.   (New) An electrochemical generator in accordance with claim 17  
2 wherein the electro-active material is mesoporous and is comprised of mesoporous beads  
3 or rods that are 5-20 micrometer (micron) and form an electrode by templated ordering,  
4 manipulative stacking or arrangement of the 5-20 micrometer (micron) mesoporous  
5 particles.

1                   35.   (New) An electrochemical generator in accordance with claim 32  
2 further comprising a separator that is a porous or mesoporous, high porosity insulating  
3 material arranged in forms of continuous layers or particle spacers, and a current  
4 collector-substrate that is selected from a group consisting of carbon, graphite paper,  
5 stainless steel, titanium or aluminum alloy and DSA.

1                   36.   (New) An electrochemical generator in accordance with claim 35  
2 wherein the insulating material is selected from a group consisting of zirconia, alumina,  
3 glass and polypropylene.

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